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# SUSTAINABLE ADDITIVES TO INCREASE THERMAL STABILITY OF DRY LUBRICANTS AND BEST PRACTICE OF USE

Dr. Thomas Morge, Tecnovo Srl; Eng. Giovanni Garoli, Koner Srl

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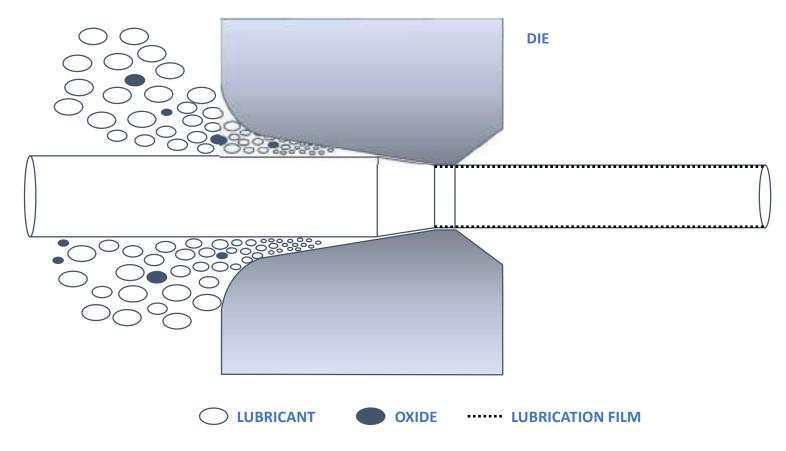
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#### **THEORY ABOUT WIRE DRAWING**



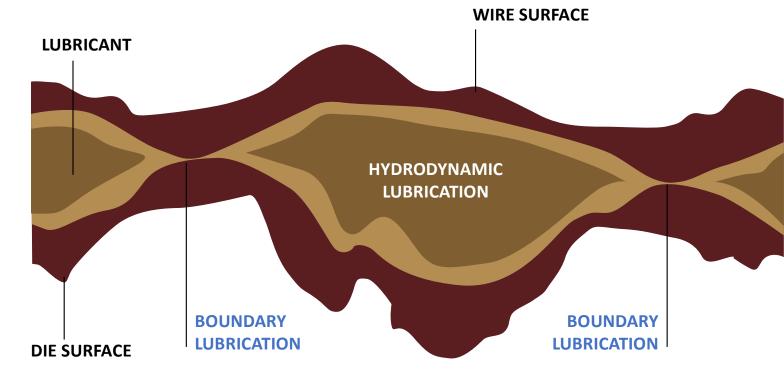






# **THEORY ABOUT WIRE DRAWING**

- Lubricant reduces friction between the surface of the wire and the die
- Hydrodynamic lubrication: thick layer, total separation between wire and die
- Boundary lubrication: thin layer between wire and die

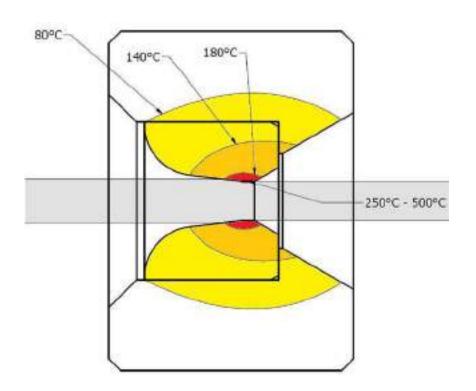








# **THEORY ABOUT WIRE DRAWING**



- Cold-forming process
- Lubricant is essential to reduce friction and wear in the contact zone, reducing temperature
- Too high temperatures can modify and embrittle the microstructure of the steel
- Lubricants must have high thermal resistance







**THERMAL RESISTANCE OF SODIUM SOAPS: SAMPLE PREPARATION** 

Sodium Soap + 3 wt.% Flame Retardant

Starting fatty acid: Stearin 70% C18, Stearic Acid 30% C16, Palmitic Acid

#### SAMPLE

Sodium Stearate (Stearin, C16 + C18) Sodium Palmitate (C16) Sodium Stearate (Pure Stearic Acid, C18) Borax Pentahydrate Sodium Metasilicate Pentahydrate Melamine Aluminium Hydroxide Sodium Phosphate Aromatic Phosphate Esther Sodium Nitrate Sodium Organic Salt + Sodium Metasilicate

Sodium Organic Salt + Sodium Metasilicate Pentahydrate







- Measure variation of heat absorbed by the material during a variation of its state
- Phase transition, melting point and oxidation
- Observed phenomena can be endothermic or exothermic

INHERT ATMOSPHERE Atmosphere: Nitrogen Ramp from 30°C to 400°C: 5°C/min Pan: Aluminium Sample weight: 10-20 mg OXIDATIVE ATMOSPHERE Atmosphere: Air Ramp from 30°C to 400°C: 5°C/min Pan: Aluminium Sample weight: 10-20 mg



*PerkinElmer DSC 8000 coupled with Intracooler 2* 







#### NITROGEN

- Inhert environment
- Cleaner graph
- Less similar to real conditions
- Better to study crystalline phases

#### AIR

- Reactive environment
- Oxidative reactions (Like burning)
- 'Noisy' graph
- Replicate real conditions

Energy involved in these transitions is the enthalpy:

#### ΔΗ=Κ Α

K is the calorimetric constant and A is the area under the curve







Phases of pure sodium soap:

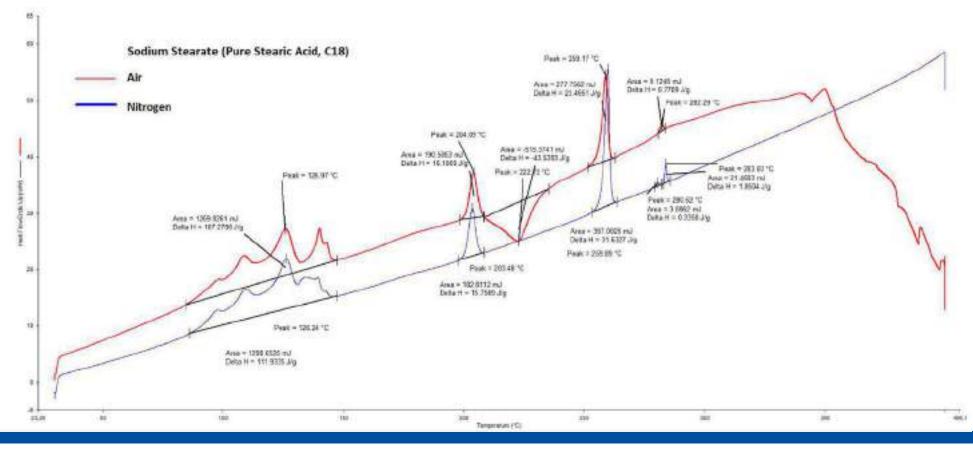
- All endothermic (requires heat)
- Each phase transition is associated to an increase of the mobility of the molecule inside the microstructure
- The first three transitions are associated to an increase of volume while in the last two the volume decreases drastically to values lower than the first one
- In AIR: oxidation at 210-225°C, exothermic. Decomposition over 300°C, exothermic

Phase change	Temperature °C
Curd fibres (omega phase) - subwaxy soap	120-126
Subwaxy soap - waxy soap	138-140
Waxy soap - subneat soap	202-206
Subneat soap - neat soap	250-259
Neat soap - isotropic liquid	283-286













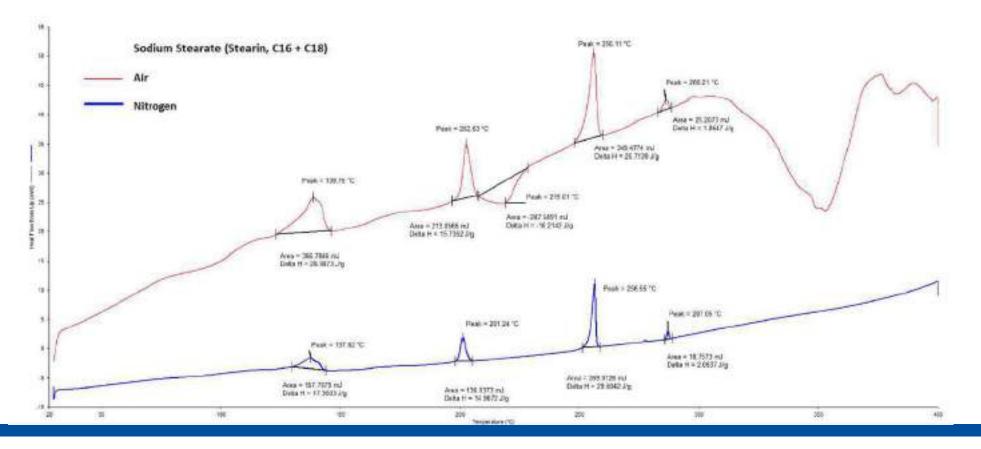


















	Sample	Peak of exothermic transition	ΔH J/g	Burning Starts	
	Sodium Stearate (Stearin, C16 + C18)	219	-22	320	
	Sodium Palmitate (C16)	218	-50	310	
	Sodium Stearate (Pure Stearic Acid, C18)	223	-41	350	
	Borax Pentahydrate	218	-35	320	
➡	Sodium Metasilicate Pentahydrate	220	-25	325	
	Melamine	11	//	315	
	Aluminium Hydroxide	212	-39	330	
➡	Sodium Phosphate	211	-41	300	
	Aromatic Phosphate Esther	//	//	310	
	Sodium Nitrate	216	-54	310	
	Sodium Organic Salt	239	-24	310	
•	Sodium Organic Salt + Sodium Metasilicate Pentahydrate	11	//	305	







- Measure variation of weight of a sample while the temperature is increased
- Thermal decomposition, oxidation, absorption, adsorption and desorption, ...

OXIDATIVE ATMOSPHERE Atmosphere: Air Ramp from 35°C to 700°C: 10°C/min Pan: Alumina Sample weight: 15-25 mg



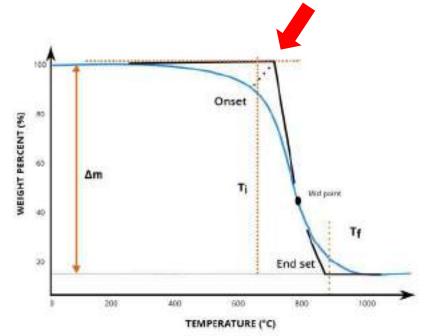
*PerkinElmer TGA 4000 coupled with PolyScience chiller* 







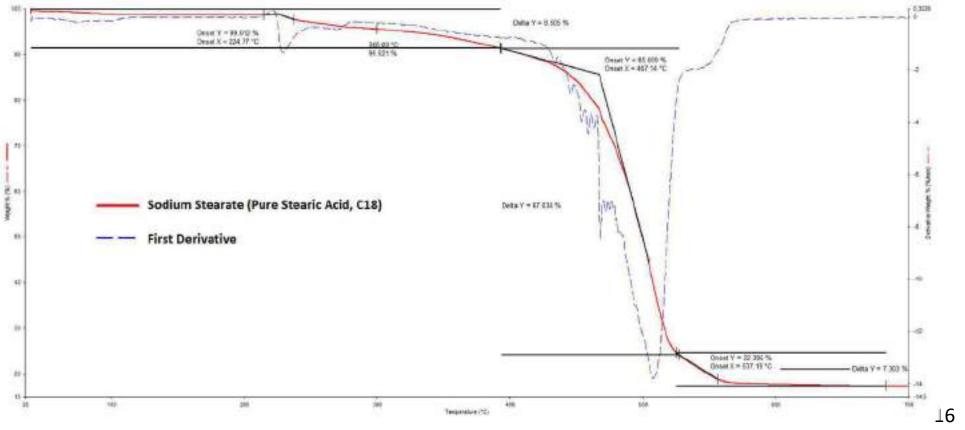
- TGA is used to evaluate the thermal stability
- The beginning of the transition is identified by the Onset temperature according to ASTM E2550.
  This is the "point in the TGA curve where a deflection is first observed from the established baseline, prior to the thermal event"
- Maximum rate of the weight loss is the inflection point of the curve and can be found through first derivative
- For soap is also useful to check the weigh loss at a fixed temperature
- Oxidative reactions produce an increase of weight







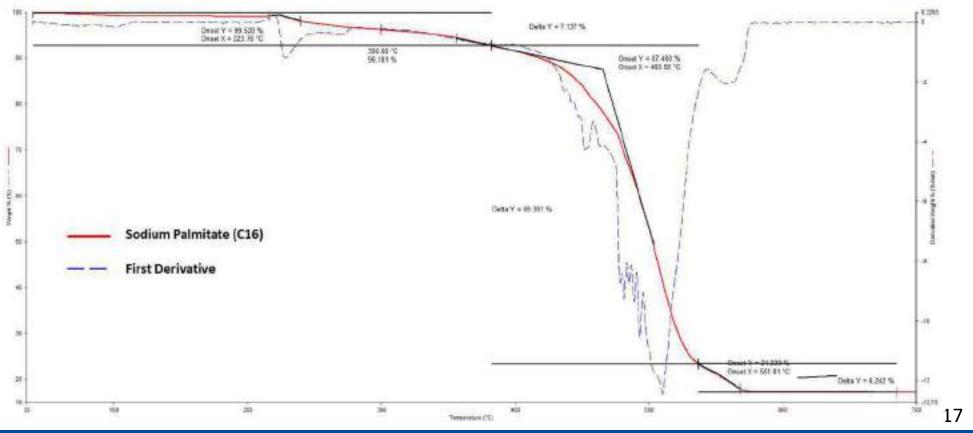








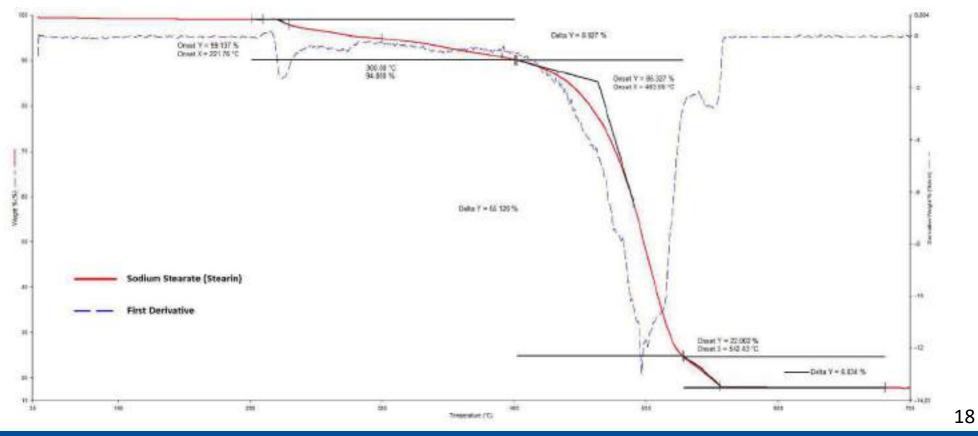


















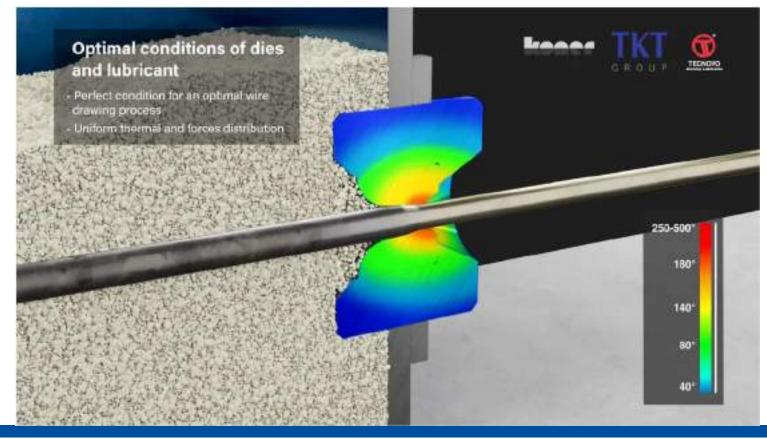
Sample	First Onset T	End	Weight loss %	Second Onset T	End	Weight loss %	Weight loss % at 300°C
Sodium Stearate (Stearin, C16 + C18)	223	425	12	425	465	83	5,5
Sodium Palmitate (C16)	221	420	12	420	463	84	3,5
Sodium Stearate (Pure Stearic Acid, C18)	225	430	11	430	467	82	4,5
Borax Pentahydrate	233	425	11,5	425	470	80	5
Sodium Metasilicate Pentahydrate	239	430	10,5	430	475	81	3,2
Melamine	224	430	15	430	473	83	4,5
Aluminium Hydroxide	222	420	13	420	460	81	6
Sodium Phosphate	230	410	9	410	450	81	3
Aromatic Phosphate Esther	245	430	14	430	461	82	4,7
Sodium Nitrate	235	380	9	425	462	81	1,8
Sodium Organic Salt	250	450	13	450	480	80	1,8
Sodium Organic Salt + Sodium Metasilicate Pentahydrate	270	435	8	435	472	79,5	1,5







**Optimal conditions of die and lubricant** 

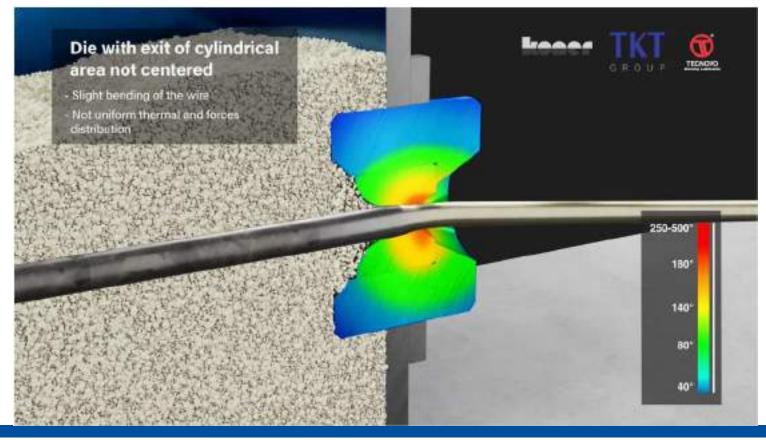








Die with exit of cylindrical area not centered

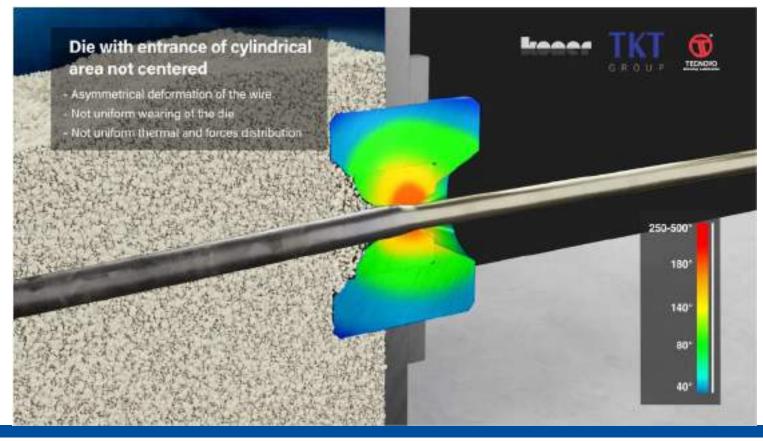








Die with entrance of cylindrical area not centered

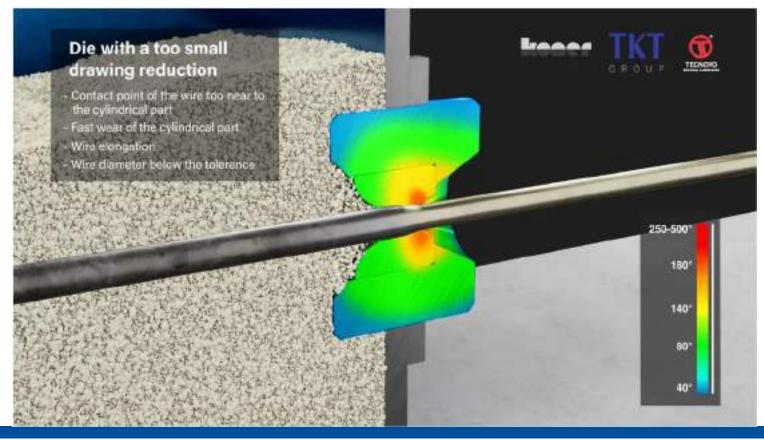








Die with a too small drawing reduction

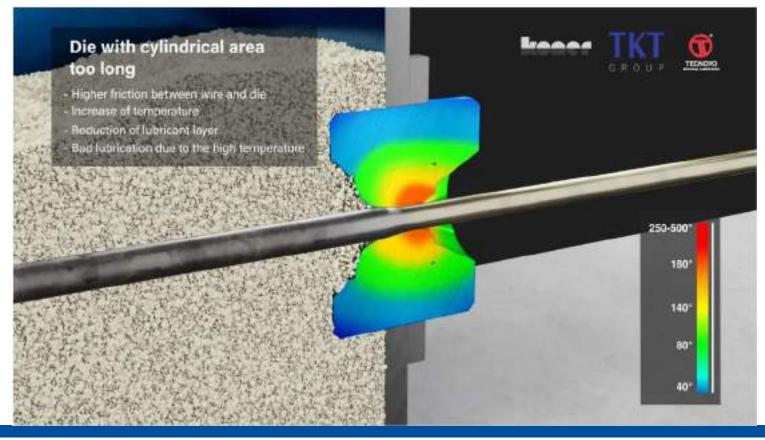








Die with cylindrical area too long

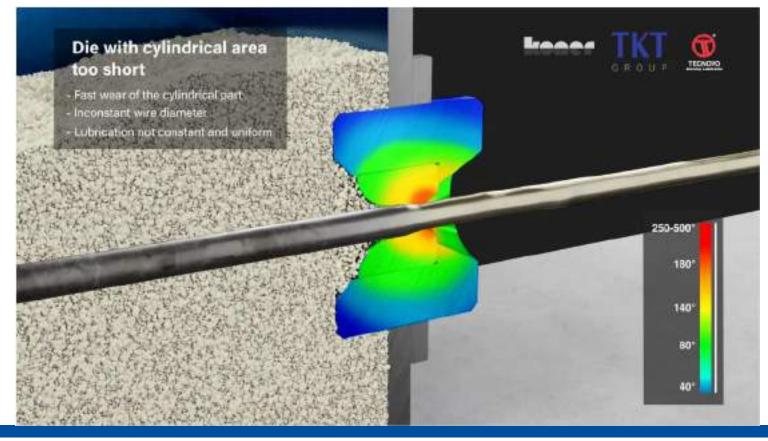








Die with cylindrical area too short

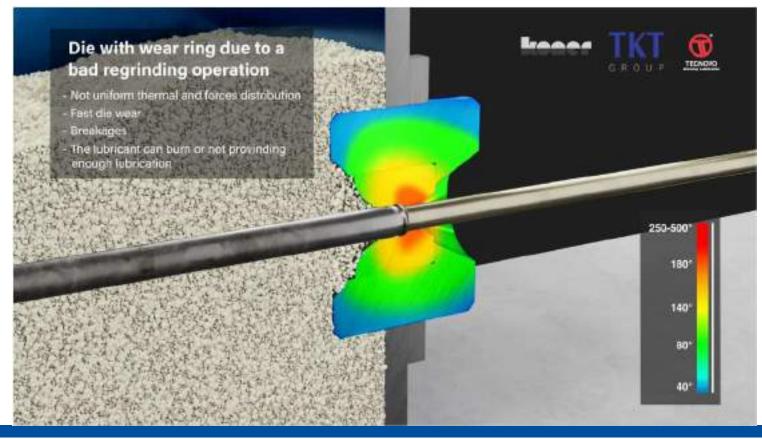








Die with wear ring due to a bad regrinding operation

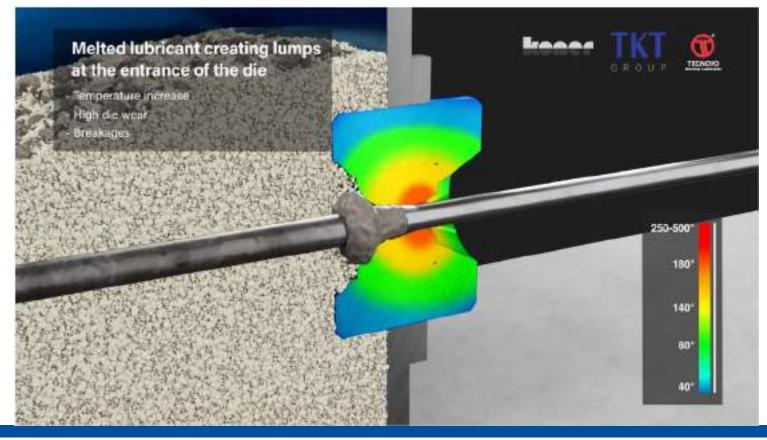








Melted lubricant creating lumps at the entrance of the die

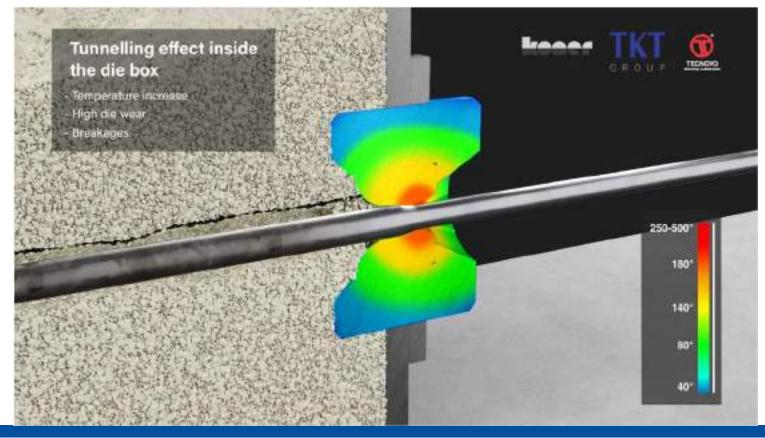








**Tunnelling effect inside the die box** 









Excess of lubricant at the die exit

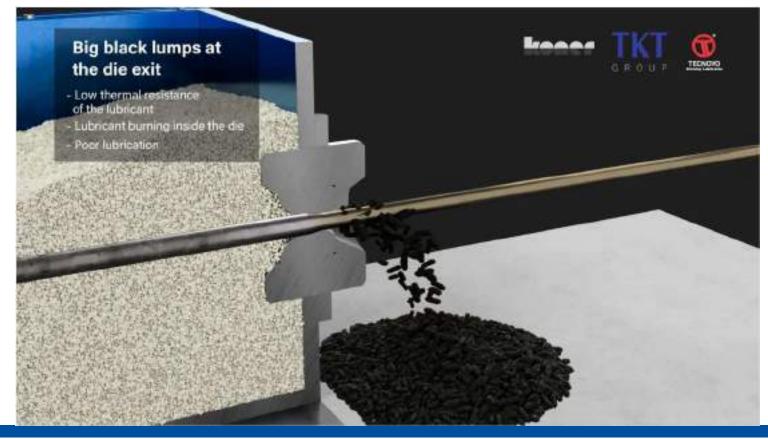








Big black lumps at the die exit









#### **CASE HISTORY: PRODUCTION OF VERY BRIGHT WIRE**









# THANK YOU FOR YOUR KIND ATTENTION



Visit us at WIRE Düsseldorf - April 15 - 19, 2024